# Synthesis of (25R)-[26-2 $\left.\mathrm{H}_{1}\right]$ Cholesterol and ${ }^{\mathbf{1}} \mathrm{H}$ N.m.r. and H.p.l.c. Resolution of (25R)- and (25S)-26-Hydroxycholesterol 

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#### Abstract

Yamogenin acetate (1) was isolated from crude diosgenin acetate and converted into (25S)-26hydroxycholesterol (6a). The absolute configuration at C-25 of (6a) was determined by $X$-ray crystallography. (25R)-[26-2 $\left.{ }^{2} \mathrm{H}_{1}\right]$ Cholesterol (10) was prepared by reduction of the 26 -tosyloxy group by $\mathrm{LiAl}^{2} \mathrm{H}_{4}$. Reverse-phase h.p.l.c. resolution without derivatization was developed for the diastereoisomers, (25R)- and (25S)-26-hydroxycholesterol. The (+)- or (-)-MTPA esters of these diastereoisomers showed distinctive ${ }^{1} \mathrm{H}$ n.m.r. signals for $26-\mathrm{H}$.


Recently we found by ${ }^{1} \mathrm{H}$ n.m.r. spectroscopy that the isopropyl methyl groups of cholesterol are magnetically inequivalent and assigned their signals by preparing stereospecifically deuteriated ( $25 S$ ) $-\left[26-{ }^{2} \mathrm{H}_{1}\right]$ cholesterol. ${ }^{1}$ Now we have prepared ( $25 R$ )[ $26^{-}{ }^{-} \mathrm{H}_{1}$ ] cholesterol (10) via ( $25 S$ ) -26 -hydroxycholesterol ( 6 ) and confirmed the n.m.r. assignments for $26-\mathrm{H}$ and $27-\mathrm{H}$ ( pro-R and pro-S methyl groups, respectively, at C-25) of cholesterol. The 26 -hydroxylated steroid is a potentially important compound as an intermediate in the biosynthesis of bile acids and probably pavoninins (shark repellent), ${ }^{2}$ as an inhibitor of cholesterol synthesis in vivo, ${ }^{3}$ and also as an intermediate in steroidal sapogenin biosynthesis in plants. ${ }^{4}$ For these studies, facile methods are essential for the preparation, resolution, and differentiation of ( $25 S$ )- and ( $25 R$ )-26-hydroxycholesterol, ( $6 a$ ) and ( 6 b). Although the ( $25 R$ )-epimer has been synthesized, ${ }^{1.5}$ the ( $25 S$ )-isomer has been less easy to prepare, Byon's method being rather complex. ${ }^{6}$

Chromatographic resolution of the diastereoisomers has been achieved in a multiple recycling process ${ }^{7}$ of the acetates of ( $6 \mathbf{a}$ ) and ( $\mathbf{6 b}$ ) and recently in a single path as the $p$-bromobenzoates of these compounds. ${ }^{8}$ Here we report a facile preparation of ( $25 S$ )-26-hydroxycholesterol (6a) from yamogenin acetate (1), which was isolated easily from crude diosgenin and a simple method for resolving (25S)- and (25R)-26-hydroxycholesterol (6a) and (6b) by reverse-phase high-performance liquid chromatography (h.p.l.c.) without derivatization. Also described is the efficient differentiation of these diastereomeric isomers by ${ }^{1} \mathrm{H}$ n.m.r. spectroscopy as their 3,26 -bis- $(+)$ - or ( - )-methoxy(trifluoromethyl)phenylacetyl esters (MTPA); only one isomer is required to determine the configuration at C-25.

## Results and Discussion

We chose yamogenin acetate (1) as a starting material for ( $25 S$ )-26-hydroxycholesterol (6a) because it has the required stereochemistry at C-25. Yamogenin is usually present in commercially available diosgenin sometimes in a concentration of up to $15 \%$. Silica gel chromatography of crude diosgenin acetate gave good resolution, affording yamogenin acetate (1). The purity was $>99.5 \%$ measured by reverse-phase h.p.l.c. after recrystallization (Figure 4 in Experimental section).
Clemmensen reduction of yamogenin acetate (1) gave (25S)$3 \beta, 16 \beta, 26$-trihydroxycholesterol (2) $(75 \%$ ), the 3 - and $26-$ hydroxy groups of which were protected selectively as $p$-nitrobenzoyl derivatives; subsequent oxidation of the 16 -hydroxy group followed by deprotection of the 3 - and 26 -positions gave the dihydroxy-16-oxo compound (5). This upon Huang-Minlon reduction afforded (25S)-26-hydroxycholesterol (6a).

(1)

(2) $R^{1}=\mathrm{H}, \mathrm{R}^{2}=\mathrm{O}_{\mathrm{H}}^{\mathrm{OH}}, \mathrm{R}^{3}=\mathrm{OH}$
(3) $\mathrm{R}^{1}=p-\mathrm{NO}_{2} \mathrm{C}_{6} \mathrm{H}_{4} \mathrm{CO}, \mathrm{R}^{2}=<_{-}^{\mathrm{OH}}, \mathrm{R}^{3}=p-\mathrm{NO}_{2} \mathrm{C}_{6} \mathrm{H}_{4} \mathrm{COO}$
(4) $R^{1}=p-\mathrm{NO}_{2} \mathrm{C}_{6} \mathrm{H}_{4} \mathrm{CO}, \mathrm{R}^{2}=\mathrm{O}, \mathrm{R}^{3}=\rho-\mathrm{NO}_{2} \mathrm{C}_{6} \mathrm{H}_{4} \mathrm{COO}$
(5) $R^{1}=H, R^{2}=0, R^{3}=O H$
(6a) $R^{1}=H, R^{2}=H_{2}, R^{3}=O H$
(7) $R^{1}=A c, R^{2}=H_{2}, R^{3}=\mathrm{OCPh}_{3}$
(8) $R^{1}=A c, R^{2}=H_{2}, R^{3}=O H$
(9) $R^{1}=\dot{A} c, R^{2}=H_{2}, R^{3}=0-p-T s$
(10) $R^{1}=H, R^{2}=H_{2}, R^{3}=D$

Although recent developments in h.p.l.c. have made it possible to separate (25S)-(6a) and (25R)-hydroxycholesterol ( $6 \mathbf{b}$ ), ${ }^{7.8}$ there subsequent identification without a pair of authentic samples is impossible. Therefore we determined the configuration at $\mathrm{C}-25$ of ( $6 \mathbf{a}$ ) as $S$ by means of $X$-ray crystallography (see Figure 1 ) $\dagger$ This ( $25 S$ )-epimer (6a) was compared by reverse-

[^0]


Figure 1. A stereoview of compound (6a)


Figure 2. Portion of $400 \mathrm{MHz}{ }^{1} \mathrm{H}$ n.m.r. spectra of $26-\mathrm{H}$ of (a) ( - )-MTPA diester of ( $25 R$ )-26-hydroxycholesterol ( 6 b ),(b) (+)-MTPA diester of ( $25 R$ )-26-hydroxycholesterol ( 6 b ), (c) ( - )-MTPA diester of ( $25 S$ )-26-hydroxycholesterol (6a), and (d) ( + )-MTPA diester of (25S)-26-hydroxycholesterol (6a) recorded on a Varian XL-400 instrument at $23^{\circ} \mathrm{C}$ in $\mathrm{CDCl}_{3}$
phase h.p.l.c. with its ( $25 R$ )-epimer, which was synthesized from diosgenin. ${ }^{1}$ As shown in Figure 5 (see Experimental section), ( $25 S$ )- ( $6 \mathbf{a}$ ) and ( $25 R$ )-26-hydroxycholesterol ( 6 b) could be reasonably resolved, without derivatization by a single passage through a TSKgel ODS-120T column developed with methanolwater ( $93: 7$ ). The ( $25 S$ )-epimer ( $6 a$ ) showed no peak due to the ( $25 R$ )-epimer ( $6 \mathbf{b}$ ). Although Clemmensen reduction of the $1,5-\mathrm{ketol}$ system has been reported to cause isomerization at $\mathrm{C}-25$ by a 1,5 -hydride shift, ${ }^{5 b}$ this was avoided in our experiment by slow addition of hydrochloric acid and also by acetal protection at C-22.
${ }^{1} \mathrm{H}$ N.m.r. spectra of the two epimers at 200 MHz were so similar that differentiation between (6a) and (6b) was difficult. However, when the two isomers were esterified with $(+)$-or ( - )-MTPA chloride, ${ }^{9}$ the hydrogens at C-26 showed diagnostic signals. As shown in Figure 2, the ( - -MTPA diester of the ( $25 R$ )-epimer ( $6 \mathbf{b}$ ) (Figure 2a) and the ( + )-MTPA diester of the (25S)-epimer (6a) (Figure 2d) show the two hydrogens at C-26 almost as a doublet signal. In contrast, these hydrogens of the


Figure 3. Portion of $400 \mathrm{MHz}{ }^{1} \mathrm{H}$ n.m.r. spectra of (a) cholesterol and (b) (25R)-[26-2 H$]$ cholesterol (10) recorded on a Varian XL-400 instrument at $23^{\circ} \mathrm{C}$ in $\mathrm{CDCl}_{3}$
( + )-MTPA diester of the ( $25 R$ )-epimer (6b) (Figure 2b) and the $(-)$-MTPA diester of the ( $25 S$ )-epimer ( $6 a$ ) (Figure 2c) appeared as a well separated pair of double doublets; even at 90 MHz n.m.r., these two epimers could be identified by their $26-\mathrm{H}$ signals. This method may be generally applicable to a determination of the configuration of a hydroxymethyl group. ${ }^{10}$ Only some signals in the ${ }^{13} \mathrm{C}$ n.m.r. spectra of (6a) and (6b) differ: C-20 to C-27, by $0.02-0.21$ p.p.m.

In a previous study, we assigned the ${ }^{1} \mathrm{H}$ n.m.r. signals of $26-\mathrm{H}$ and $27-\mathrm{H}$ of cholesterol using ( $25 S$ ) - $\left[26-{ }^{2} \mathrm{H}_{1}\right]$ cholesterol. ${ }^{1}$ In order to confirm the assignments ( $25 S$ )-26-hydroxycholesterol (6a) was transformed to (25R)-[26- $\left.{ }^{2} \mathrm{H}_{1}\right]$ cholesterol (10) according to a known method. ${ }^{1}$ As shown in Figure 3, the ${ }^{1} \mathrm{H}$ n.m.r. spectrum of (10) shows $27-\mathrm{H}$ as a doublet $\left(J_{\text {H.H }} 6.5 \mathrm{~Hz}\right)$ at $\delta_{\mathrm{H}}$ 0.865 and $26-\mathrm{H}$ as a doublet of triplets at $\delta_{\mathrm{H}} 0.842$ which couples to $25-\mathrm{H}\left(J_{\mathrm{H}, \mathrm{H}} 6.5 \mathrm{~Hz}\right)$ and $26-{ }^{2} \mathrm{H}\left(J_{\mathrm{H} .{ }^{2} \mathrm{H}} 1.8 \mathrm{~Hz}\right)$. This confirms the signal assignments at $\delta_{\mathrm{H}} 0.862$ and 0.866 to $26-\mathrm{H}$ (pro- $R$ methyl group at $\mathrm{C}-25$ ) and $27-\mathrm{H}$ (pro- $S$ methyl group at $\mathrm{C}-25$ ) of cholesterol, respectively. In the ${ }^{13} \mathrm{C}$ n.m.r. spectrum of compound (10), C-26 appeared at $\delta_{\mathrm{C}} 22.26$ as a triplet ( ${ }^{1} J_{\mathrm{C}},{ }^{2} \mathrm{H} 19 \mathrm{~Hz}$, ${ }^{1} \Delta \delta_{\left.\mathrm{C}_{( }^{2} \mathrm{H}\right)}-0.3$ p.p.m.) and $\mathrm{C}-27$ at $\delta_{\mathrm{C}} 22.80$ as a singlet, which agrees with the assignments ${ }^{11}$ originally made by Popják on the basis of biosynthetic findings. ${ }^{12}$ We therefore conclude that our findings confirm a si-face reduction of the $24(25)$-double


Figure 4. H.p.l.c. analysis of (a) yamogenin acetate (1), (b) a mixture of yamogenin and diosgenin, and (c) diosgenin. Conditions: column, TSK gel ODS-120T $250 \times 4 \mathrm{~mm}$ i.d.; solvent, $7 \%$ water in methanol, 1 $\mathrm{ml} / \mathrm{min}$; detector, UVILOG-5IIIA at 208 nm
bond of lanosterol in the biosynthesis of cholesterol proposed by Caspi, ${ }^{13}$ because the $Z$-methyl group (C-27) at C-25 of lanosterol originating from $\mathrm{C}-3^{\prime}$ of mevalonate turns into one of the methyl groups at $\mathrm{C}-25$ of cholesterol, which appeared at $\delta_{\mathrm{C}} 22.9 .1^{12}$ We confirmed the signal to correspond to $\mathrm{C}-27$ (pro-S methyl group at C-25).

## Experimental

M.p.s were taken on a hot plate and are uncorrected. ${ }^{1} \mathrm{H}$ N.m.r. spectra were recorded on a Varian XL-200 ( 200.057 MHz ) or an XL-400 ( 399.948 MHz ) spectrometer in $\left[{ }^{2} \mathrm{H}\right]$ chloroform unless otherwise stated using a $5-\mathrm{mm}$ spinning tube. ${ }^{13} \mathrm{C}$ N.m.r. spectra were determined on a Varian XL-100-12A spectrometer operating at 25.16 MHz in $\left[{ }^{2} \mathrm{H}\right]$ chloroform using a $10-\mathrm{mm}$ spinning spherical tube. Chemical shifts are given in p.p.m. downfield from internal tetramethylsilane. Acquisition times of ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ measurements were 5 and 1.6 s , respectively. Optical rotation was determined on a Perkin-Elmer 241 Polarimeter and mass spectra were recorded on a Hitachi RMU-8GN spectrometer. H.p.l.c. was performed with a Waters 600 multisolvent delivery system equipped with a U6K injector and UVILOG5IIIA u.v. detector.

Isolation of Yamogenin Acetate (1).-Diosgenin (22 g; purchased from Wako Pure Chemical Industries, Ltd.) was acetylated with acetic anhydride ( 80 ml ) and pyridine ( 60 ml ) to obtain the crude acetate ( 24 g ). Chromatography of the acetate $(1 \mathrm{~g})$ on silica gel [ $2 \times$ Lobar B, hexane-chloroform-ethyl acetate (15:1:1), $2.5 \mathrm{ml} / \mathrm{min}]$ gave diosgenin acetate ( $200-250$
ml fraction, 917 mg ) and raw yamogenin acetate ( $250-300 \mathrm{ml}$ fraction, 83 mg ), which when recrystallized from methanol, gave yamogenin acetate (1), m.p. $180-182^{\circ} \mathrm{C},[x]_{\mathrm{D}}{ }^{24}-130.8^{\circ}(c$ $0.655 \mathrm{CHCl}_{3}$ ); $\delta_{\mathrm{H}}\left(\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N}\right) 0.844(3 \mathrm{H}, \mathrm{s}, 18-\mathrm{H}), 0.975(3 \mathrm{H}, \mathrm{s}$, $19-\mathrm{H}), 1.086(3 \mathrm{H}, \mathrm{d}, J 7.2 \mathrm{~Hz}, 27-\mathrm{H}), 1.161(3 \mathrm{H}, \mathrm{d}, J 6.8 \mathrm{~Hz}$, $21-\mathrm{H}), 2.069(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 3.38(1 \mathrm{H}, \mathrm{d}, J 11 \mathrm{~Hz}, 26 \beta-\mathrm{H}), 4.07(1 \mathrm{H}$, dd, $J 11$ and $3 \mathrm{~Hz}, 26 x-\mathrm{H}), 4.52(1 \mathrm{H}, \mathrm{dt}, J 8$ and $7 \mathrm{~Hz}, 16-\mathrm{H})$, $4.82(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H})$, and $5.35(1 \mathrm{H}, \mathrm{d}, J 5 \mathrm{~Hz}, 6-\mathrm{H}) ; \delta_{\mathrm{c}} 37.01(\mathrm{C}-1)$, 27.78 (C-2), 73.92 (C-3), 38.14 (C-4), 139.75 (C-5), 122.38 (C-6), 32.08 (C-7), 31.45 (C-8), 50.02 (C-9), 36.77 (C-10), 20.86 (C-11), 39.79 (C-12), 40.27 (C-13), 56.48 (C-14), 31.83 (C-15), 80.92 (C-16), 62.01 (C-17), 16.30 (C-18), 19.35 (C-19), 42.18 (C-20), 14.36 (C-21), 109.75 (C-22), 26.02 (C-23), 25.83 (C-24), 27.12 (C-25), 65.14 ( $\mathrm{C}-26$ ), 16.08 ( $\mathrm{C}-27$ ), and 21.40 and 170.5 $\left(\mathrm{CH}_{3} \mathrm{CO}\right)$. Reverse-phase h.p.l.c. is shown in Figure 4.
(25S)-Cholest-5-ene-3 $\beta, 16 \beta, 26$-triol (2).-Zinc amalgam [freshly prepared from zinc powder ( 30 g ) and $\mathrm{HgCl}_{2}(3 \mathrm{~g})$ ] was added to yamogenin acetate ( 1 g ) dissolved in $90 \%$ ethanol ( 100 ml ). Concentrated hydrochloric acid ( 30 ml ) was then added over 2.5 h to the above mixture whilst it was stirred vigorously and heated under reflux; the heating was then continued for a further 30 min . The mixture was cooled and the inorganic material filtered off. The filtrate was concentrated under reduced pressure to 30 ml and extracted with chloroform ( 200 ml ); the extract was then washed successively with water ( 100 ml ), $3 \%$ aqueous sodium hydrogen carbonate ( 100 ml ), and water ( 100 ml ) and evaporated. Chromatography of the residue ( 1 g ) on silica gel eluted with hexane-chloroform-ethyl acetateisopropyl alcohol (7:7:7:1) gave ( $25 S$ )-cholest-5-ene$3 \beta, 16 \beta, 26-$ triol (2) ( 721 mg ), m.p. $177-178{ }^{\circ} \mathrm{C}$ (from methanol), $[x]_{\mathrm{D}}{ }^{23}-26.5^{\circ}$ (c $1.02 \mathrm{CHCl}_{3}-\mathrm{MeOH}, 3: 1$ ) (Found: C, 76.9; $\mathrm{H}, 11.0 \% ; M^{+}, 418 . \mathrm{C}_{2}, \mathrm{H}_{46} \mathrm{O}_{3}$ requires C, $77.5 ; \mathrm{H}, 11.1 \% ; M$, 418); $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}-\mathrm{CD}_{3} \mathrm{OD} 3: 1\right.$ ) $0.887(3 \mathrm{H}, \mathrm{s}, 18-\mathrm{H}), 0.914(3 \mathrm{H}$, d, $J 6.7 \mathrm{~Hz}, 27-\mathrm{H}$ ), $0.982(3 \mathrm{H}, \mathrm{d}, J 6.7 \mathrm{~Hz}, 21-\mathrm{H}$ ), $1.022(3 \mathrm{H}, \mathrm{s}$, $19-\mathrm{H}), 3.37$ and $3.43(2 \mathrm{H}, \mathrm{AB}$ part of ABX, $J 10.5,6.5$, and 6 Hz , $26-\mathrm{H}), 3.4(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 4.36(1 \mathrm{H}, \mathrm{m}, 16-\mathrm{H})$, and $5.36(1 \mathrm{H}, \mathrm{d}$, $J 5 \mathrm{~Hz}, 6-\mathrm{H}) ; \delta_{\mathrm{c}} 37.38(\mathrm{C}-1), 31.19(\mathrm{C}-2), 72.36(\mathrm{C}-3), 42.94$ (C-4), 142.20 (C-5), 122.24 (C-6), 32.21 (C-7), 32.75 (C-8), 51.58 (C-9), 36.92 (C-10), 21.80 (C-11), 38.40 (C-12), 43.26 (C-13), 55.69 (C-14), 41.22 (C-15), 72.77 (C-16), 62.73 (C-17), 13.46 (C-18), 19.86 (C-19), 37.61 (C-20), 18.80 (C-21), 34.84 (C-22), 24.92 (C-23), 32.90 (C-24), 37.91 (C-25), 68.40 (C-26), and 17.32 (C-27).
(25S)-3 3,26 -Bis-p-nitrobenzoyloxycholest-5-en-16ß-ol (3).-$p$-Nitrobenzoyl chloride ( 529 mg ) was added to the triol (2) ( 596 mg ) dissolved in dried dichloromethane ( 27 ml ) and pyridine $(170 \mathrm{mg})$ at $0^{\circ} \mathrm{C}$. The mixture was stirred at room temperature for 2 h . Methanol ( 2 ml ) was added to destroy the excess of chloride and the solution was stirred at room temperature for 20 min , and then diluted with chloroform ( 100 ml ), washed successively with water, $1 \%$ hydrochloric acid, and water, and evaporated. Chromatography on silica gel eluted with hexane-chloroform-ethyl acetate ( $7: 2: 2$ ) gave ( $25 S$ ) $-3 \beta, 26-$ bis-p-nitrobenzoyloxycholest-5-en-16ß-ol (3) ( 499 mg ), m.p. $190-190.5^{\circ} \mathrm{C}$ (from methanol), $[x]_{\mathrm{D}}{ }^{23} 0.0^{\circ}$ (c $1.009 \mathrm{CHCl}_{3}+1$ drop of MeOH ) (Found: C, 68.5; H, 7.4. Calc. for $\mathrm{C}_{41} \mathrm{H}_{52} \mathrm{~N}_{2} \mathrm{O}_{9}$ : $\mathrm{C}, 68.7 ; \mathrm{H}, 7.3) ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}+1\right.$ drop of $\left.\mathrm{CD}_{3} \mathrm{OD}\right) 0.909(3 \mathrm{H}, \mathrm{s}$, $18-\mathrm{H}), 0.999(3 \mathrm{H}, \mathrm{d}, J 6.4 \mathrm{~Hz}, 21-\mathrm{H}$ ), $1.045(3 \mathrm{H}, \mathrm{d}, J 6.7 \mathrm{~Hz}, 27-$ H), 1.093 ( $3 \mathrm{H}, \mathrm{s}, 19-\mathrm{H}$ ), $2.50(2 \mathrm{H}, \mathrm{d}, J 7 \mathrm{~Hz}, 4-\mathrm{H}$ ), 4.19 and 4.30 ( $2 \mathrm{H}, \mathrm{AB}$ part of ABX, $J 10.5,7$, and $6 \mathrm{~Hz}, 26-\mathrm{H}$ ), $4.35(1 \mathrm{H}, \mathrm{m}$, $16-\mathrm{H}), 4.92(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 5.45(1 \mathrm{H}, \mathrm{d}, J 5 \mathrm{~Hz}, 6-\mathrm{H}), 8.21$ and 8.28 $\left(4 \mathrm{H}, \mathrm{A}_{2} \mathrm{~B}_{2}, J 9 \mathrm{~Hz}, \mathrm{ArH}\right)$, and 8.21 and $8.29\left(4 \mathrm{H}, \mathrm{A}_{2} \mathrm{~B}_{2}, J 9 \mathrm{~Hz}\right.$, ArH).
(25S)-3,26-Bis-p-nitrobenzoyloxycholest-5-en-16-one (4).Jones' reagent ( 1.0 ml ) was added to bis-p-nitrobenzoate (3)


Figure 5. H.p.l.c. analysis of (a) (25S)-26-h ydroxycholesterol (6a), (b) a mixture of (6a) and (6b), and (c) (25R)-26-hydroxycholesterol (6b). Conditions: column, TSKgel ODS-120T $250 \times 4 \mathrm{~mm}$ i.d.; solvent, $7 \%$ water in $\mathrm{MeOH}, 1 \mathrm{ml} / \mathrm{min}$; detector, UVILOG-5IIIA at 205 nm
( 635 mg ) in acetone ( 40 ml ) at $0^{\circ} \mathrm{C}$ and stirred at $0^{\circ} \mathrm{C}$ for 30 min . The reaction mixture was diluted with water ( 100 ml ) and extracted with dichloromethane ( 150 ml ), and the extract washed with water, and evaporated to give ( $25 S$ ) $-3 \beta, 26$-bis-p-nitrobenzoyloxycholest-5-en-16-one (4) ( 624 mg ), m.p. 157$159{ }^{\circ} \mathrm{C}$ (from methanol), $[\alpha]_{\mathrm{D}}{ }^{23.5}-80.6{ }^{\circ} \mathrm{C}\left(c 1.021 \mathrm{CHCl}_{3}\right)$ (Found: C, 68.7; H, 7.0; N, 4.0. Calc. for $\mathrm{C}_{41} \mathrm{H}_{50} \mathrm{~N}_{2} \mathrm{O}_{9}$ : C, 68.9 ; H, $7.1 ; \mathrm{N}, 4.0) ; \delta_{\mathrm{H}} 0.846(3 \mathrm{H}, \mathrm{s}, 18-\mathrm{H}), 0.990(3 \mathrm{H}, \mathrm{d}, J 6.2 \mathrm{~Hz}$, $21-\mathrm{H}), 1.042(3 \mathrm{H}, \mathrm{d}, J 6.8 \mathrm{~Hz}, 27-\mathrm{H}), 1.109(3 \mathrm{H}, \mathrm{s}, 19-\mathrm{H}), 2.31$ $(2 \mathrm{H}, \mathrm{d}, J 6.6 \mathrm{~Hz}, 4-\mathrm{H}), 4.17$ and $4.28(2 \mathrm{H}, \mathrm{AB}$ part of ABX, $J 10.5,6.5$, and $6.0 \mathrm{~Hz}, 26-\mathrm{H}), 4.92(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 5.45(1 \mathrm{H}, \mathrm{d}$, $J 5 \mathrm{~Hz}, 6-\mathrm{H})$, and 8.21 and $8.28\left(4 \mathrm{H}, \mathrm{A}_{2} \mathrm{~B}_{2}, J 9 \mathrm{~Hz}, \mathrm{ArH}\right)$, and 8.21 and $8.29\left(4 \mathrm{H}, \mathrm{A}_{2} \mathrm{~B}_{2}, J 9 \mathrm{~Hz}, \mathrm{ArH}\right)$.
(25S)-3 $\beta, 26$-Dihydroxycholest-5-en-16-one (5).-A solution of 16 -oxo-bis-p-nitrobenzoate (4) ( 500 mg ) in $1 \%$ methanolic potassium hydroxide ( 40 ml ) was refluxed for 50 min . The reaction mixture was diluted with water $(150 \mathrm{ml})$ and extracted with chloroform ( 400 ml ), and the extract washed with water ( 300 ml ) and evaporated to give ( $25 S$ )-3 3,26 -dihydroxycholest-5-en-16-one (5) ( 280 mg ), m.p. $166-167.5^{\circ} \mathrm{C}$ (from methanol), $[x]_{\mathrm{D}}{ }^{23.5}-171.6^{\circ}\left(c 0.87 \mathrm{CHCl}_{3}\right)$ (Found: C, 77.3; H, 10.6. Calc. for $\left.\mathrm{C}_{27} \mathrm{H}_{44} \mathrm{O}_{3}: \mathrm{C}, 77.8 ; \mathrm{H}, 10.7\right) ; \delta_{\mathrm{H}} 0.837(3 \mathrm{H}, \mathrm{s}, 18-\mathrm{H}), 0.924(3$ $\mathrm{H}, \mathrm{d}, J 6.7 \mathrm{~Hz}, 27-\mathrm{H}), 0.976(3 \mathrm{H}, \mathrm{d}, J 6.5 \mathrm{~Hz}, 21-\mathrm{H}), 1.040(3 \mathrm{H}, \mathrm{s}$, $19-\mathrm{H}), 3.40$ and $3.52(2 \mathrm{H}, \mathrm{AB}$ part of ABX, $J 12,6.5$, and 6 Hz , $26-\mathrm{H}), 3.5(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H})$, and $5.36(1 \mathrm{H}, \mathrm{d}, J 5 \mathrm{~Hz}, 6-\mathrm{H})$.
(25S)-Cholest-5-en-26-ol (6a).-A solution of the 16-oxo diol (5) $(199 \mathrm{mg})$ in triethylene glycol ( 10.4 ml$)$ was heated at $130^{\circ} \mathrm{C}$ for 1.5 h with hydrazine hydrochloride ( 360 mg ) and $80 \%$ hydrazine hydrate ( 1.8 g ). Potassium hydroxide ( 900 mg ) was
added to the mixture which was then heated at $210^{\circ} \mathrm{C}$ for 3 h with distillation of the water. The mixture was cooled, diluted with water ( 100 ml ), and extracted with chloroform ( 450 ml ). The extract was washed with water ( 100 ml ) and then evaporated. Purification of the residue ( 187 mg ) by reverse-phase h.p.l.c. (Figure 5) gave ( $25 S$ )-cholest-5-en-26-ol ( $6 a$ ) ( 141 mg ), m.p. $177-178{ }^{\circ} \mathrm{C}$ (from methanol) (lit., ${ }^{6} 171-174{ }^{\circ} \mathrm{C}$ ). $[x]_{\mathrm{D}}{ }^{23}$ $-44.0^{\circ}\left(c \quad 1.00 \mathrm{CHCl}_{3}\right.$ ) (Found: C, $80.7 ; \mathrm{H}, 11.6 \% ; M^{+}, 402$. $\mathrm{C}_{27} \mathrm{H}_{46} \mathrm{O}_{2}$ requires C, $\left.80.5 ; \mathrm{H}, 11.5 \% ; M, 402\right) ; \delta_{\mathrm{H}} 0.679(3 \mathrm{H}, \mathrm{s}$, $18-\mathrm{H}), 0.920(6 \mathrm{H}, \mathrm{d}, J 6.5 \mathrm{~Hz}, 21-\mathrm{H}$ and $27-\mathrm{H}), 1.009(3 \mathrm{H}, \mathrm{s}$, $19-\mathrm{H}), 3.43$ and $3.52(2 \mathrm{H}, \mathrm{AB}$ part of ABX $J 10.5,6.5$, and 6 Hz , $26-\mathrm{H}), 3.50(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H})$, and $5.36(1 \mathrm{H}, \mathrm{d}, J 5 \mathrm{~Hz}, 6-\mathrm{H}) ; \delta_{\mathrm{C}} 37.30$ (C-1), 31.71 (C-2), 71.82 (C-3), 42.36 (C-4), 140.82 (C-5), 121.70 (C-6), 31.94 ( $\mathrm{C}-7$ and $\mathrm{C}-8$ ), 50.19 (C-9), 36.54 (C-10), 21.12 (C-11), 39.83 (C-12), 42.36 (C-13), 56.81 (C-14), 24.31 (C-15), 28.26 (C-16), 56.18 (C-17), 11.89 (C-18), 19.41 (C-19), 35.85 (C-20*), 18.77 (C-21), 36.30 (C-22), 23.50 (C-23), 33.71 (C-24), 35.80 (C-25*), 68.37 (C-26), and 16.74 (C-27). (*These assignments may be reversed.)

X-Ray Structure Determination of (6a).-Crystals were grown from ethyl acetate solution by slow evaporation.

Crystal data. $\mathrm{C}_{27} \mathrm{H}_{46} \mathrm{O}_{2}, M=402.7$. Orthorhombic, $a=$ 11.887(1), $b=32.635(3), c=6.297(1) \AA, V=2442.8(4) \AA^{3}$ (by least-squares refinement of diffractometer angles for 24 centred reflections, $20<\theta<28^{\circ}, \lambda=1.54178 \AA$ ), space group $P 2_{1} 2_{1} 2_{1}, Z=4, D_{x}=1.10 \mathrm{~g} \mathrm{~cm}^{-3}$; colourless needles; crystal dimensions $0.10 \times 0.15 \times 0.40 \mathrm{~mm}, \mu\left(\mathrm{Cu}-K_{\alpha}\right)=0.51 \mathrm{~mm}^{-1}$.

Data collection and processing. Rigaku AFC-5 diffractometer, $\omega / 2 \theta$ scan, graphite-monochromated $\mathrm{Cu}-K_{\alpha}, 2420$ unique reflections measured $\left(\theta \leqq 65.5^{\circ},+h, k, l\right)$, no absorption correction, giving 2236 with $I>\sigma(I)$.

Structure analysis and refinement. H Atoms were located from difference map by direct methods. ${ }^{14}$ Block diagonal leastsquares refinement was carried out with anisotropic temperature factors for non- H atoms and isotropic for H atoms; there was no correction for secondary extinction. $\Sigma \omega \Delta^{2}$ Minimized, $\Delta=\left|F_{\mathrm{o}}\right|-\left|F_{\mathrm{c}}\right|, \omega=1 / \sigma^{2}\left(F_{0}\right)$ for $\left|F_{\mathrm{c}}\right|>2 \sigma\left(F_{0}\right), \omega=0$ for $\left|F_{\mathrm{c}}\right| \leqq 2 \sigma\left(F_{0}\right)$ or $|\Delta|>5 \sigma\left(F_{0}\right), \quad \sigma\left(F_{\mathrm{o}}\right)=\left[\sigma_{1}{ }^{2}\left(F_{0}\right)+0.00247\right.$ $\left.\left|F_{0}\right|^{2}\right]^{1 / 2}, \sigma_{1}\left(F_{0}\right)=$ e.s.d. based on counting errors; ${ }^{15} R=0.058$, $R_{\mathrm{w}}=0.080$. Atomic scattering factors from ref. 16. Calculations performed on a FACOM-M340R computer. Atomic co-ordinates are given in the Table. Bond lengths and angles are anisotropic and isotropic temperature factors, are available on request from the Cambridge Crystallographic Data Centre. $\dagger$
(+)- and (-)-MTPA Esters of (25S)- and (25R)-26-Hydroxycholesterol ( $\mathbf{6 a}$ ) and ( $\mathbf{6 b}$ ).-A solution of ( $25 S$ )-26-hydroxycholesterol ( $6 \mathbf{6}$ ) $(5 \mathrm{mg})$ and $(+)$-MTPA chloride ${ }^{9}$ ( 3 drops) in pyridine-dichloromethane ( $1: 1$ ) ( 1 ml ) was left at room temperature for 15 h . Water was added to destroy the excess of chloride after which the product was extracted with ether ( 30 ml ) and the extract evaporated. Chromatography of the residue on silica gel t.l.c. (hexane-chloroform-ethyl acetate, 15:1:1) gave the (+)-MTPA diester of (6a) ( 5 mg ), $\delta_{\mathrm{H}} 0.666(3 \mathrm{H}, \mathrm{s}$, $18-\mathrm{H}), 0.895(3 \mathrm{H}, \mathrm{d}, J 6.7 \mathrm{~Hz}, 21-\mathrm{H}), 0.921(3 \mathrm{H}, \mathrm{d}, J 6.4 \mathrm{~Hz}$, $27-\mathrm{H}), 1.003(3 \mathrm{H}, \mathrm{s}, 19-\mathrm{H}), 3.553\left(3 \mathrm{H}, \mathrm{q},{ }^{5} J_{\mathrm{H}, \mathrm{F}} 1.2 \mathrm{~Hz}, \mathrm{OMe}\right)$, $3.566\left(3 \mathrm{H}, \mathrm{q},{ }^{5} J_{\mathrm{H} . \mathrm{F}} 1.2 \mathrm{~Hz}\right.$, OMe $), 4.157(2 \mathrm{H}, \mathrm{d}, J 6.0 \mathrm{~Hz}, 26-\mathrm{H})$, $4.88(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 5.40(1 \mathrm{H}, \mathrm{d}, J 4 \mathrm{~Hz}, 6-\mathrm{H}), 7.40(6 \mathrm{H}, \mathrm{m}$, ArH ), and 7.53 ( $4 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$ ).

Treatment of ( $6 \mathbf{a}$ ) with ( - )-MTPA using the same procedure as above yielded the ( - )-MTPA diester of ( 6 a ) $(5 \mathrm{mg}), \delta_{\mathrm{H}} 0.667$ ( $3 \mathrm{H}, \mathrm{s}, 18-\mathrm{H}$ ), 0.898 ( $3 \mathrm{H}, \mathrm{d}, J 6.5 \mathrm{~Hz}, 21-\mathrm{H}$ ), $0.913(3 \mathrm{H}, \mathrm{d}, J 6.4$ $\mathrm{Hz}, 27-\mathrm{H}), 1.002(3 \mathrm{H}, \mathrm{s}, 19-\mathrm{H}), 3.553\left(3 \mathrm{H}, \mathrm{q},{ }^{5} J_{\mathrm{H} . \mathrm{F}} 1.2 \mathrm{~Hz}, \mathrm{OMe}\right)$,

[^1]Table. Fractional atomic co-ordinates for compound (6a) with e.s.d.s in parentheses

|  | $x$ | $y$ | $=$ |
| :--- | :---: | :---: | ---: |
| $\mathrm{C}(1)$ | $0.2190(3)$ | $0.1475(1)$ | $0.4613(5)$ |
| $\mathrm{C}(2)$ | $0.2082(3)$ | $0.1018(1)$ | $0.5242(6)$ |
| $\mathrm{C}(3)$ | $0.3016(3)$ | $0.0900(1)$ | $0.6754(5)$ |
| $\mathrm{C}(4)$ | $0.2948(3)$ | $0.1162(1)$ | $0.8749(5)$ |
| $\mathrm{C}(5)$ | $0.2994(2)$ | $0.1615(1)$ | $0.8217(4)$ |
| $\mathrm{C}(6)$ | $0.3716(2)$ | $0.1854(1)$ | $0.9204(5)$ |
| $\mathrm{C}(7)$ | $0.3836(2)$ | $0.2310(1)$ | $0.8861(5)$ |
| $\mathrm{C}(8)$ | $0.2836(2)$ | $0.2489(1)$ | $0.7633(5)$ |
| $\mathrm{C}(9)$ | $0.2523(2)$ | $0.2204(1)$ | $0.5764(4)$ |
| $\mathrm{C}(10)$ | $0.2166(2)$ | $0.1767(1)$ | $0.6542(4)$ |
| $\mathrm{C}(11)$ | $0.1658(2)$ | $0.2397(1)$ | $0.4252(5)$ |
| $\mathrm{C}(12)$ | $0.1946(3)$ | $0.2843(1)$ | $0.3579(5)$ |
| $\mathrm{C}(13)$ | $0.2144(2)$ | $0.3116(1)$ | $0.5524(4)$ |
| $\mathrm{C}(14)$ | $0.3108(2)$ | $0.2914(1)$ | $0.6783(4)$ |
| $\mathrm{C}(15)$ | $0.3505(3)$ | $0.3239(1)$ | $0.8376(5)$ |
| $\mathrm{C}(16)$ | $0.3335(3)$ | $0.3643(1)$ | $0.7159(5)$ |
| $\mathrm{C}(17)$ | $0.2670(2)$ | $0.3549(1)$ | $0.5082(4)$ |
| $\mathrm{C}(18)$ | $0.1073(2)$ | $0.3159(1)$ | $0.6833(5)$ |
| $\mathrm{C}(19)$ | $0.0979(2)$ | $0.1770(1)$ | $0.7516(7)$ |
| $\mathrm{C}(20)$ | $0.1863(3)$ | $0.3897(1)$ | $0.4438(6)$ |
| $\mathrm{C}(21)$ | $0.1270(3)$ | $0.3826(1)$ | $0.2340(7)$ |
| $\mathrm{C}(22)$ | $0.2454(3)$ | $0.4321(1)$ | $0.4411(6)$ |
| $\mathrm{C}(23)$ | $0.3499(3)$ | $0.4361(1)$ | $0.3074(7)$ |
| $\mathrm{C}(24)$ | $0.3900(3)$ | $0.4811(1)$ | $0.3059(7)$ |
| $\mathrm{C}(25)$ | $0.5110(3)$ | $0.4880(1)$ | $0.2350(6)$ |
| $\mathrm{C}(26)$ | $0.5318(3)$ | $0.4729(1)$ | $0.0129(7)$ |
| $\mathrm{C}(27)$ | $0.5435(3)$ | $0.5326(1)$ | $0.2630(9)$ |
| $\mathrm{O}(3)$ | $0.2984(2)$ | $0.0470(1)$ | $0.7272(4)$ |
| $\mathrm{O}(26)$ | $0.6468(2)$ | $0.4740(1)$ | $-0.0438(5)$ |
|  |  |  |  |

$3.568\left(3 \mathrm{H}, \mathrm{q},{ }^{5} J_{\mathrm{H} . \mathrm{F}} 1.2 \mathrm{~Hz}, \mathrm{OMe}\right), 4.070(1 \mathrm{H}, \mathrm{dd}, J 10.4$ and 7.2 $\left.\mathrm{Hz}, 26-\mathrm{H}_{\mathrm{a}}\right), 4.239\left(1 \mathrm{H}, \mathrm{dd}, J 10.4\right.$ and $\left.5.6 \mathrm{~Hz}, 26-\mathrm{H}_{\mathrm{b}}\right), 4.88(1 \mathrm{H}$, $\mathrm{m}, 3-\mathrm{H}), 5.42(1 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}, 6-\mathrm{H}), 7.40(6 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$, and 7.53 ( $4 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$ ).

Treatment of (6b) with (+)-MTPA using the same procedure as above yielded the $(+)$-MTPA diester of ( $\mathbf{6 b}$ ) $(5 \mathrm{mg}), \delta_{H} 0.666$ ( $3 \mathrm{H}, \mathrm{s}, 18-\mathrm{H}$ ), $0.899(3 \mathrm{H}, \mathrm{d}, J 6.5 \mathrm{~Hz}, 21-\mathrm{H}), 0.910(3 \mathrm{H}, \mathrm{d}, J 6.7$ $\mathrm{Hz}, 27-\mathrm{H}), 1.003(3 \mathrm{H}, \mathrm{s}, 19-\mathrm{H}), 3.553\left(3 \mathrm{H}, \mathrm{q},{ }^{5} J_{\mathrm{H} . \mathrm{F}} 1.2 \mathrm{~Hz}, \mathrm{OMe}\right)$, $3.566\left(3 \mathrm{H}, \mathrm{q},{ }^{5} J_{\text {H.F }} 1.2 \mathrm{~Hz}, \mathrm{OMe}\right), 4.072(1 \mathrm{H}, \mathrm{dd}, J 10.4$ and $\left.7.2 \mathrm{~Hz}, 26-\mathrm{H}_{\mathrm{a}}\right), 4.230\left(1 \mathrm{H}, \mathrm{dd}, J 10.4\right.$ and $\left.5.6 \mathrm{~Hz}, 26-\mathrm{H}_{\mathrm{b}}\right), 4.88$ ( $1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}$ ), $5.403(1 \mathrm{H}, \mathrm{d}, J 4 \mathrm{~Hz}, 6-\mathrm{H}), 7.40(6 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$, and $7.53(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$.

Treatment of (6b) with (-)-MTPA using the above procedure yielded the ( - )-MTPA diester of $(6 \mathrm{~b})(5 \mathrm{mg}), \delta_{\mathrm{H}}$ $0.666(3 \mathrm{H}, \mathrm{s}, 18-\mathrm{H}), 0.899(3 \mathrm{H}, \mathrm{d}, J 6.8 \mathrm{~Hz}, 21-\mathrm{H}), 0.916(3 \mathrm{H}, \mathrm{d}$, $J 6.8 \mathrm{~Hz}, 27-\mathrm{H}), 1.002(3 \mathrm{H}, \mathrm{s}, 19-\mathrm{H}), 3.553\left(3 \mathrm{H}, \mathrm{q},{ }^{5} J_{\mathrm{H} . \mathrm{F}} 1.2 \mathrm{~Hz}\right.$, OMe), $3.568\left(3 \mathrm{H}, \mathrm{q},{ }^{5} J_{\mathrm{H} . F} 1.2 \mathrm{~Hz}, \mathrm{OMe}\right), 4.154(1 \mathrm{H}, \mathrm{dd}, J 11$ and $\left.5.5 \mathrm{~Hz}, 26-\mathrm{H}_{\mathrm{a}}\right), 4.155\left(1 \mathrm{H}, \mathrm{dd}, J 11\right.$ and $\left.6.5 \mathrm{~Hz}, 26-\mathrm{H}_{\mathrm{b}}\right), 4.88(1 \mathrm{H}$, $\mathrm{m}, 3-\mathrm{H}), 5.42(1 \mathrm{H}, \mathrm{d}, J 5 \mathrm{~Hz}, 6-\mathrm{H}), 7.53$ and $7.40(6 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$, and $7.53(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$.
(25S)-3及-Acetoxy-26-trityloxycholest-5-ene (7).-Trityl chloride ( 134 mg ) was added to the diol ( $6 a$ ) $(97.5 \mathrm{mg}$ ) dissolved in dried pyridine ( 1.2 ml ) and the mixture was stirred at $80^{\circ} \mathrm{C}$ for 2 h ; acetic anhydride ( 0.8 ml ) was then added and the mixture stirred at $80^{\circ} \mathrm{C}$ for a further 30 min . Methanol ( 3 ml ) was added to the reaction mixture to destroy the excess of reagents and then ice-water ( 50 ml ) was added. The product was extracted with ethyl acetate ( 210 ml ), washed successively with $0.5 \%$ hydrochloric acid ( 120 ml ), water ( 120 ml ), $1 \%$ sodium carbonate ( 120 ml ), and water ( 120 ml ), and then evaporated. Chromatography of the residue ( 300 mg ) on silica gel (hexane-chloroform-ethyl acetate, $18: 1: 1$ ) gave ( $25 S$ )-3-acetoxy-26-trityloxycholest-5-ene (7) (118 mg) (together with a
small amount of 3,26-bis-tritylate), $m / z 626\left(M^{+}\right), \delta_{H} 0.665(3 \mathrm{H}$, $\mathrm{s}, 18-\mathrm{H}), 0.865(3 \mathrm{H}, \mathrm{d}, J 6.5 \mathrm{~Hz}, 21-\mathrm{H}), 0.950(3 \mathrm{H}, \mathrm{d}, J 6.7 \mathrm{~Hz}$, $27-\mathrm{H}), 1.018(3 \mathrm{H}, \mathrm{s}, 19-\mathrm{H}), 2.031(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 2.85$ and $2.96(2 \mathrm{H}$, AB part of ABX, $J 9,6.5$, and $6 \mathrm{~Hz}, 26-\mathrm{H}), 4.61(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H})$, $5.38(1 \mathrm{H}, \mathrm{d}, J 5 \mathrm{~Hz}, 6-\mathrm{H})$, and $7.2-7.5(15 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$.
(25S)-3ß-Acetoxycholest-5-en-26-ol (8).-A solution of 26-Otritylate (7) ( 115 mg , without further purification) in ethanol ( 7 ml ) and hydrochloric acid ( 0.3 ml ) was stirred at room temperature for 2 h . The reaction mixture was diluted with water ( 70 ml ), extracted with ethyl acetate, and the extract washed successively with water, $1 \%$ aqueous sodium carbonate, and water, and then evaporated. Chromatography of the residue on silica gel with hexane-chloroform-ethyl acetate ( $2: 1: 1$ ) as eluant gave ( $25 S$ )-3-acetox ycholest-5-en-26-ol (8) ( 58 mg ), m.p. $132-133{ }^{\circ} \mathrm{C}$ (from methanol), $[x]_{\mathrm{D}}{ }^{25}-46.6^{\circ}\left(c 1.00 \mathrm{CHCl}_{3}\right)$ (Found: C, 78.1; H, 10.7. Calc. for $\mathrm{C}_{29} \mathrm{H}_{48} \mathrm{O}_{3}$ : C, 78.3; H, 10.9); $\delta_{\mathrm{H}} 0.676(3 \mathrm{H}, \mathrm{s}, 18-\mathrm{H}), 0.922(6 \mathrm{H}, \mathrm{d}, J 6.5 \mathrm{~Hz}, 21-\mathrm{H}$ and $27-\mathrm{H})$, $1.018(3 \mathrm{H}, \mathrm{s}, 19-\mathrm{H}), 2.032(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 3.42$ and $3.52(2 \mathrm{H}, \mathrm{AB}$ part of ABX, $J 10.5,6.5$, and $6 \mathrm{~Hz}, 26-\mathrm{H}), 4.60(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H})$, and $5.37(1 \mathrm{H}, \mathrm{d}, J 5 \mathrm{~Hz}, 6-\mathrm{H}) ; \delta_{\mathrm{C}} 37.01(\mathrm{C}-1), 27.78$ (C-2), $73.99(\mathrm{C}-3)$, 38.13 (C-4), 139.69 (C-5), 122.62 (C-6), 29.72 (C-7), 31.88 (C-8), 50.06 (C-9), 36.60 (C-10), 21.03 (C-11), 39.75 (C-12), 42.34 (C13), 56.70 (C-14), 24.27 (C-15), 28.23 (C-16), 56.12 (C-17), 11.86 (C-18), 19.30 (C-19), 35.85 (C-20*), 18.74 (C-21), 36.27 (C-22), 23.48 (C-23), 33.69 (C-24), 35.78 (C-25*), 68.36 (C-26), 16.71 (C27), and 21.40 and $170.51\left(\mathrm{CH}_{3} \mathrm{CO}\right)$. (* These assignments may be reversed.)
(25S)-3-Acetoxy-26-tosyloxycholest-5-ene (9).—A mixture of the monohydroxy acetate (8) ( 30 mg ) and toluene-p-sulphonyl chloride ( 46 mg ) in dried pyridine $(0.8 \mathrm{ml})$ was stirred at room temperature for 5 h after which water was added. The product was extracted with ether ( 210 ml ) and the extract washed successively with $0.3 \%$ hydrochloric acid ( 150 ml ), water ( 50 ml ), $1 \%$ aqueous sodium carbonate ( 50 ml ), and water ( 150 ml ), and then evaporated. Chromatography of the residue ( 51 mg ) on silica gel with hexane-chloroform-ethyl acetate ( $8: 1: 1$ ) as eluant gave ( $25 S$ )-3-acetoxy-26-tosyloxycholest-5-ene (9) (40 $\mathrm{mg}) ; \delta_{\mathrm{H}} 0.665(3 \mathrm{H}, \mathrm{s}, 18-\mathrm{H}), 0.871(3 \mathrm{H}, \mathrm{d}, J 6.2 \mathrm{~Hz}, 21-\mathrm{H}), 0.881$ ( $3 \mathrm{H}, \mathrm{d}, J 6.5 \mathrm{~Hz}, 27-\mathrm{H}$ ), 1.018 ( $3 \mathrm{H}, \mathrm{s}, 19-\mathrm{H}$ ), $2.032(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac})$, $2.33(2 \mathrm{H}, \mathrm{d}, J 8 \mathrm{~Hz}, 4-\mathrm{H}), 2.450(3 \mathrm{H}, \mathrm{s}, \mathrm{ArMe}), 3.81$ and 3.89 ( $2 \mathrm{H}, \mathrm{AB}$ part of ABX, $J 9,6.5$, and $6 \mathrm{~Hz}, 26-\mathrm{H}$ ), $4.61(1 \mathrm{H}, \mathrm{m}$, $3-\mathrm{H}), 5.38(1 \mathrm{H}, \mathrm{d}, J 5 \mathrm{~Hz}, 6-\mathrm{H})$, and 7.36 and $7.80\left(4 \mathrm{H}, \mathrm{A}_{2} \mathrm{~B}_{2}\right.$, $J 8 \mathrm{~Hz}, \mathrm{ArH})$.
(25R)-[26-2 $\left.\mathrm{H}_{1}\right]$ Cholesterol (10).--Lithium aluminium tetradeuteride $\left(\mathrm{LiAl}^{2} \mathrm{H}_{4}\right)(80 \mathrm{mg})$ was added to the $26-O$-tosylate (9) $(40 \mathrm{mg})$ in dried ether $(4 \mathrm{ml})$ and stirred at room temperature for 20 h . The excess of reagent was destroyed with $0.5 \%$ aqueous sodium hydrogen carbonate and the mixture filtered. The filtrate was then diluted with ether ( 300 ml ), washed with water $(300 \mathrm{ml})$, and then evaporated. Chromatography of the residue $(20 \mathrm{mg})$ on silica gel with hexane-chloroform-ethyl acetate ( $4: 1: 1$ ) as eluant gave crude $(25 R)-\left[26-{ }^{2} \mathrm{H}\right]$ cholesterol ( 20 mg ), which was purified with reverse-phase h.p.l.c. (column TSKgel-ODS-120T $250 \times 20 \mathrm{~mm}$ i.d. developed with methanol equipped with a refractometric detector), m.p. $148-148.5^{\circ} \mathrm{C}$ (from methanol), $[x]_{\mathrm{D}}{ }^{23.5}-40.8^{\circ}$ (c $0.77 \mathrm{CHCl}_{3}$ ); m/z 387 $\left(M^{+}+{ }^{1} \mathrm{H} 99 \%\right), 386(1 \%) ; \delta_{\mathrm{H}} 0.679(3 \mathrm{H}, \mathrm{s}, 18-\mathrm{H}), 0.842(2 \mathrm{H}$, $\mathrm{dt}, J_{\mathrm{H} . \mathrm{H}} 6.5$ and $\mathrm{J}_{\mathrm{H} .{ }^{2} \mathrm{H}} 1.8 \mathrm{~Hz},{ }^{1} \Delta \delta_{2_{\mathrm{H}}}-0.020$ p.p.m. $\left.26-\mathrm{H}\right), 0.865$ $(3 \mathrm{H}, \mathrm{d}, J 6.6 \mathrm{~Hz}, 27-\mathrm{H}), 0.915(3 \mathrm{H}, \mathrm{d}, J 6.6 \mathrm{~Hz}, 21-\mathrm{H}), 1.009(3 \mathrm{H}$, $\mathrm{s}, 19-\mathrm{H}), 3.525(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H})$, and $5.353(1 \mathrm{H}, \mathrm{d}, J 5.5 \mathrm{~Hz}, 6-\mathrm{H}) ; \delta_{\mathrm{C}}$ 37.27 (C-1), 31.69 (C-2), 71.82 (C-3), 42.32 (C-4), 140.76 (C-5), 121.72 (C-6), 31.92 (C-7 and C-8), 50.15 (C-9), 36.52 (C-10), 21.10 (C-11), 39.80 (C-12), 42.32 (C-13), 56.78 (C-14), 24.30 (C-15), 28.24 (C-16), 56.16 (C-17), 11.87 (C-18), 19.41 (C-19), 35.79 (C-20), 18.73 (C-21), 36.20 (C-22), 23.83 (C-23), 39.50
(C-24), $27.94\left({ }^{2} \Delta \delta_{\left.\mathrm{C}^{2} \mathrm{H}\right)}-0.06\right.$ p.p.m. C-25), $22.26\left(\mathrm{t},{ }^{1} J_{\mathrm{C},{ }^{2} \mathrm{H}} 19 \mathrm{~Hz}\right.$, ${ }^{1} \Delta \delta_{\left.\mathrm{C}^{2} \mathrm{H}\right)}-0.3$ p.p.m., C-26), and 22.80 (C-27).

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[^0]:    $\dagger$ Yamogenin (1) has the same nuclear configuration as that of its (25R)epimer, diosgenin, since (1) and diosgenin are chemically correlated; the absolute configuration of diosgenin was determined by $\boldsymbol{X}$-ray crystallography (W. Kline and J. Buckingham, 'Atlas of Stereochemistry,' Chapman and Hall, London, 1974, p. 126).

[^1]:    $\dagger$ See Instructions to Authors (1987), J. Chem. Soc., Perkin Trans. 1, 1987, Issue 1.

